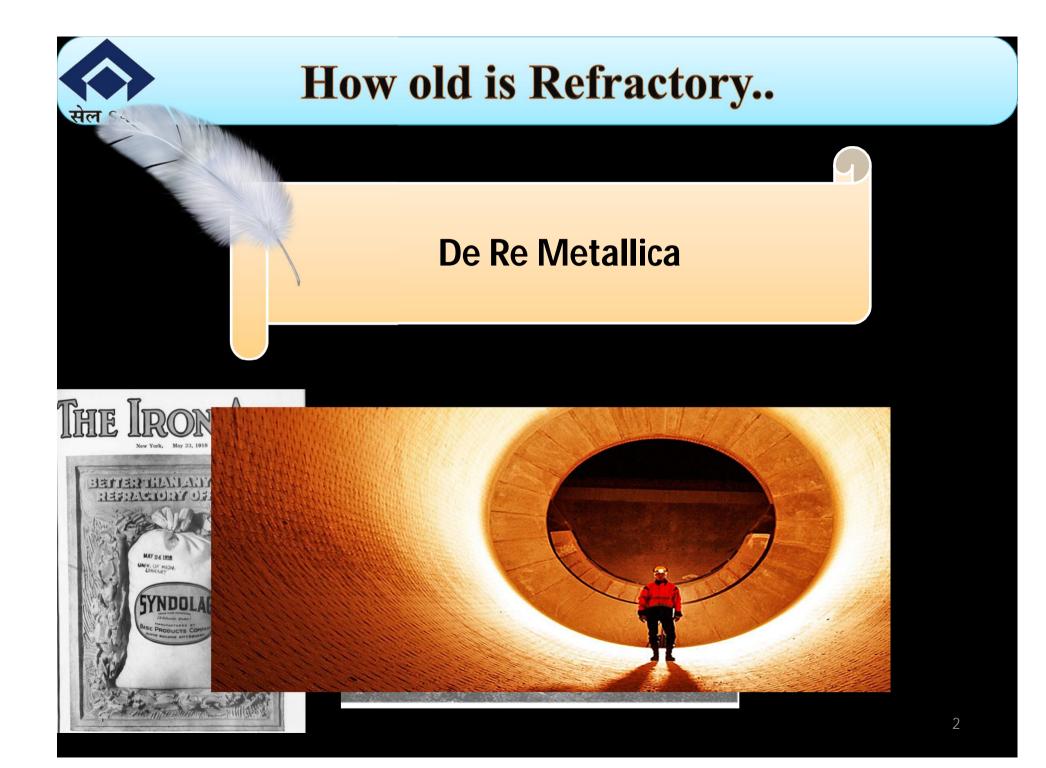
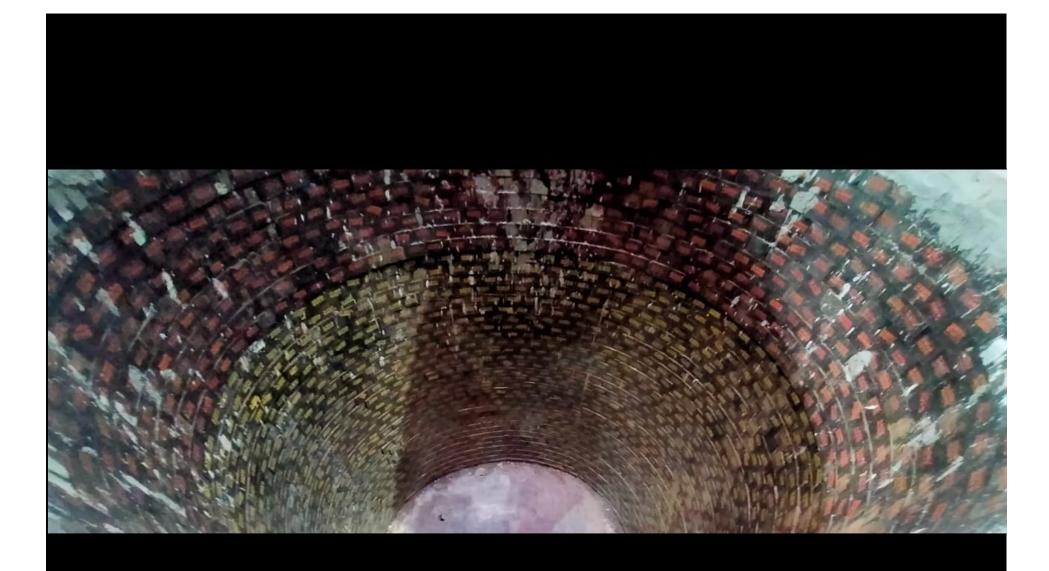
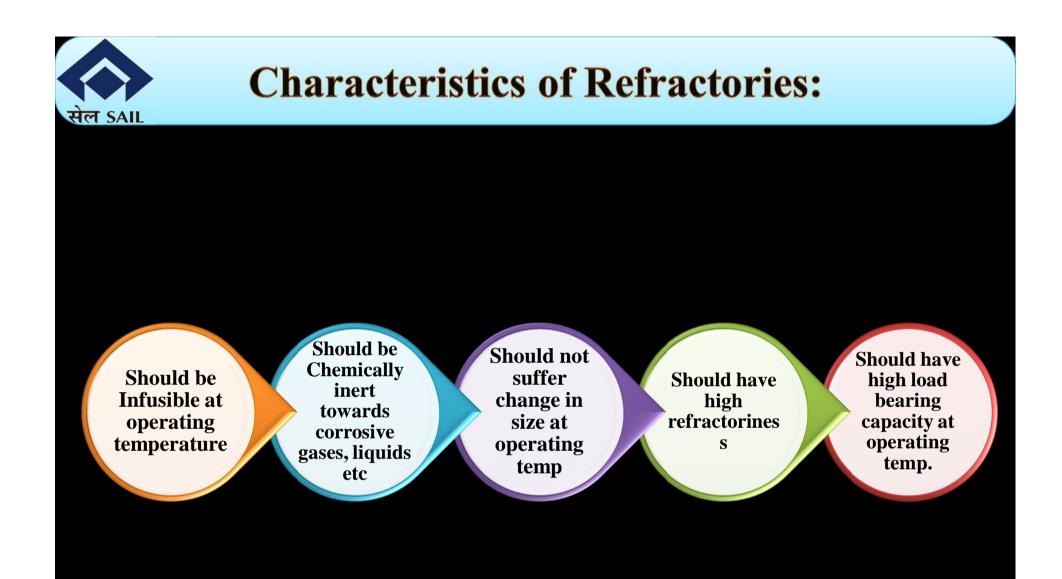
STUDY ON RECYCLING OF USED MAG-CARBON BRICK IN PRODUCTION OF MAG-CARBON BRICKS

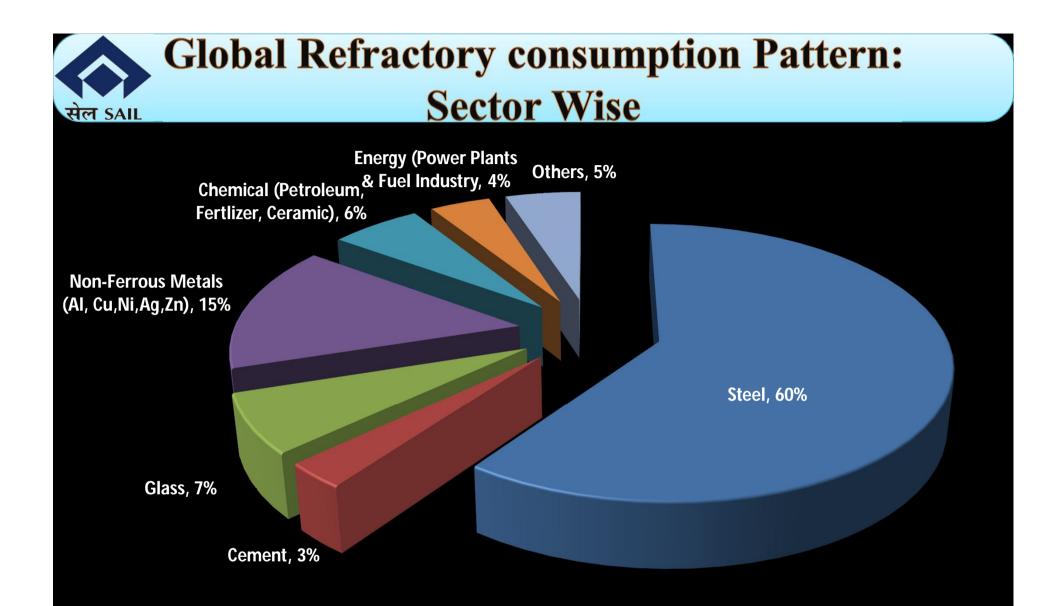


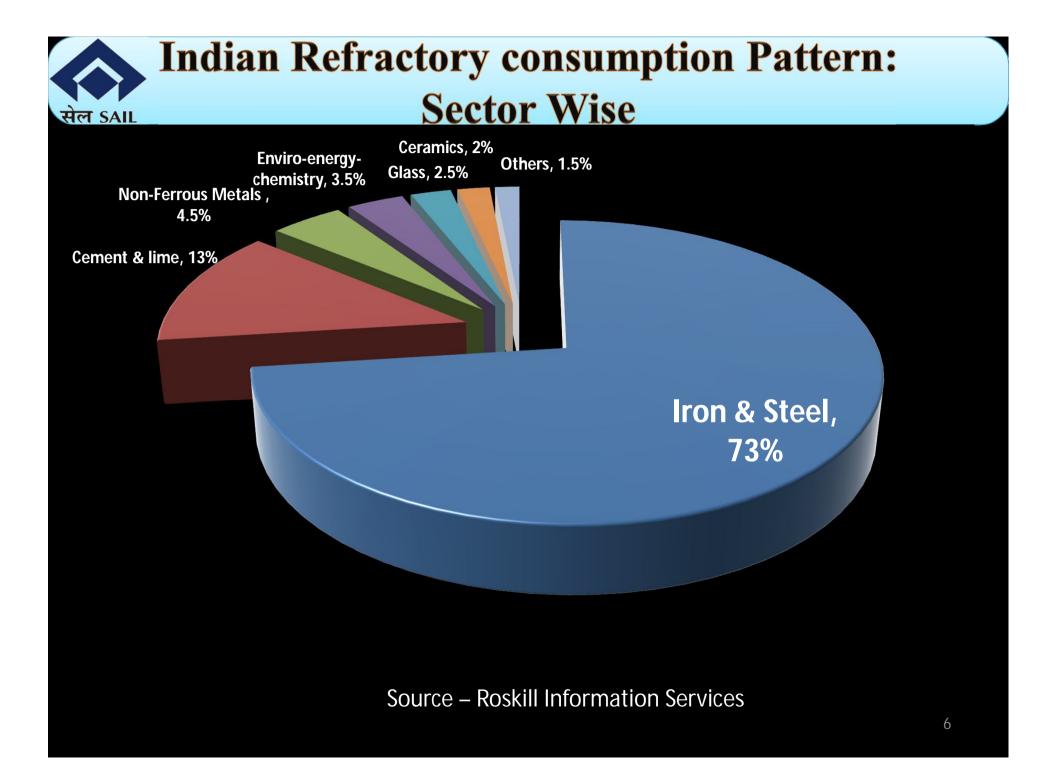
Pallavi Singh, Sr.Mgr. (R&C) SAIL Refractory Unit Bhilai

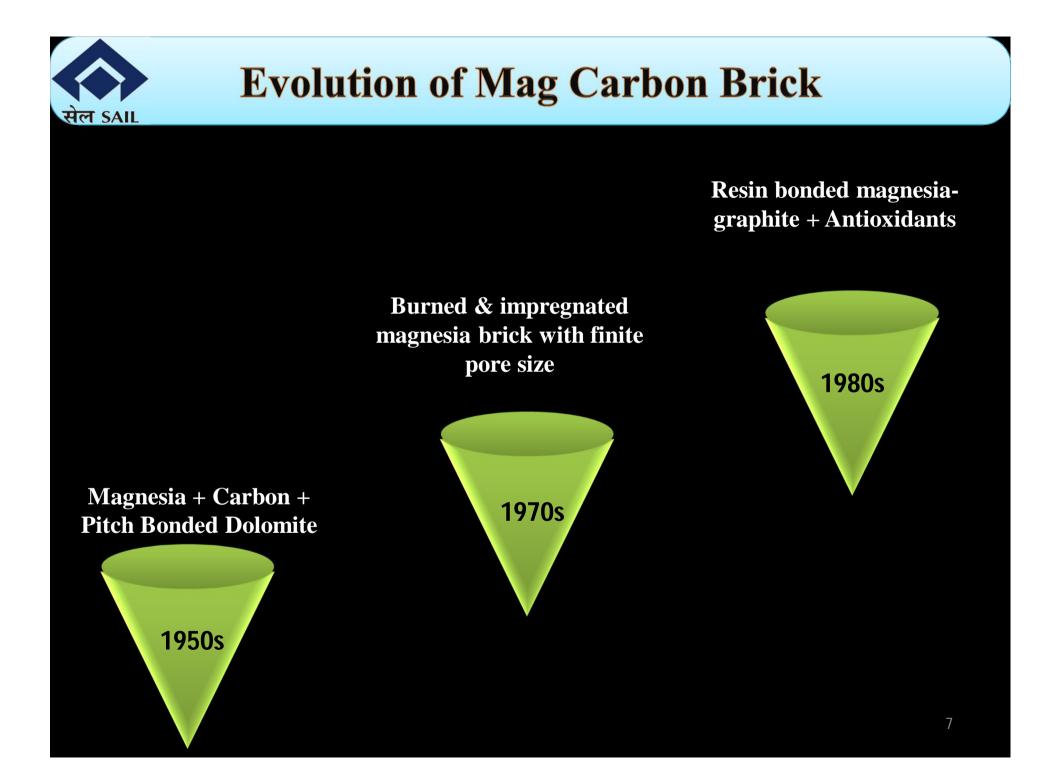


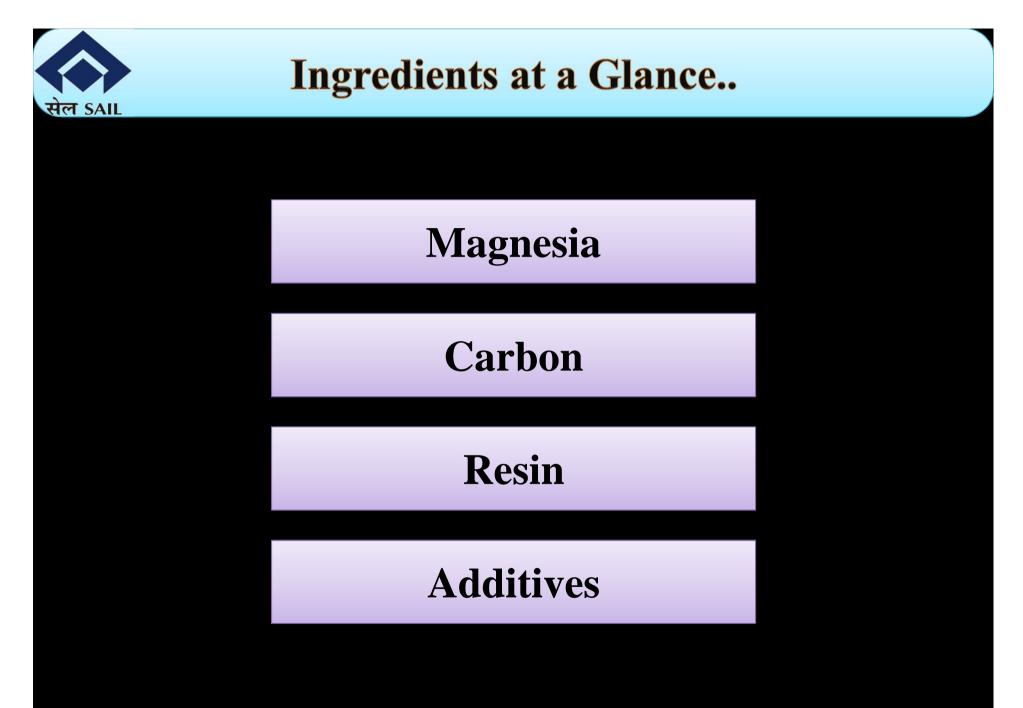














Classification of Magnesia..

Based On Origin

Fused Magnesia

Sea Water Magnesia

Dead Burnt Magnesia



More About Magnesia..

	MgO%	CaO%	Fe ₂ O ₃ %	Al ₂ O ₃ %	SiO ₂ %	B ₂ O ₃ %	BD (in gm/cc)	Crystal size (in Micron)	CaO/SiO ₂
Fused Magnesia	98	1	0.25	0.15	0.5		3.5	400	2
Synthetic Magnesia	98.35	0.83	0.57	0.12	0.2	0.02	3.4	60	4
Dead Burnt Magnesia (Australian origin)	96	2.7	0.5	0.2	0.8	0.002	3.4	100	>2.5
Sea Water Magnesia	97	2.4	0.25	0.12	0.35	0.03	3.4	110	>5
Dead Burnt Magnesia(Brazilian origin)	98.33	0.87	0.39	0.07	0.25	negligibl e	3.31	122	3.5



Fused Magnesia

 Fused Magnesia is produced by fusion of Raw Magnesite or Calcined magnesia by means of electric arcing and cooled down to crystallization.

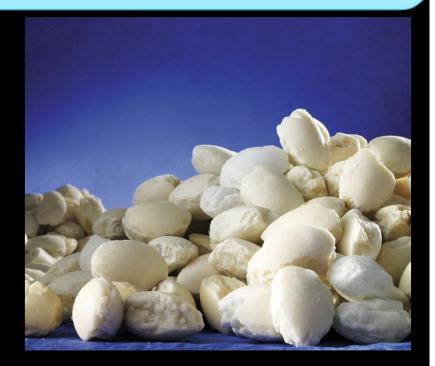


- It has advantages of high purity and large crystal size.
- Thermal shock resistance is very poor for this material.



Sea Water Magnesia

 Sea Water Magnesia is produced by extraction of Magnesium Chloride salt from sea water followed by beneficiation, calcination and high temperature firing.



• This material is very pure with cryptocrystalline structure and has high CaO:SiO₂ ratio.



Dead Burnt Magnesia

 Natural deposit of Raw Magnesite(MgCO₃) is first beneficiated then calcined, followed by high temperature firing, to produce Dead Burnt Magnesia.

 This material contains negligible amount of Boron with high lime silica ratio.





Carbon in the form of Graphite

Flaky graphite is mainly used because of its excellent physical properties such as:



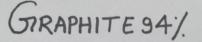
The non wettability of the slag

High thermal conductivity

Low thermal expansibility

In addition, graphite do not fuse with refractories at high temperature.

The purity and the flake size of graphite has a great influence on the performance of magnesia carbon bricks







Resin as Binder..

It contains high quantity of fixed carbon which gives strong bonding property,

It possesses high dry strength because of its thermosetting nature,

It produces less hazardous gas than tar / pitch,

At curing temperature (around 200^oC), it polymerizes which gives isotropic interlocking structure, and

Cold crushing strength (CCS) increases with the increase of resin content.



Other Additives..



Carbon components contained in MgO-C bricks are oxidized by oxygen in the atmosphere or by iron oxide in the slag.

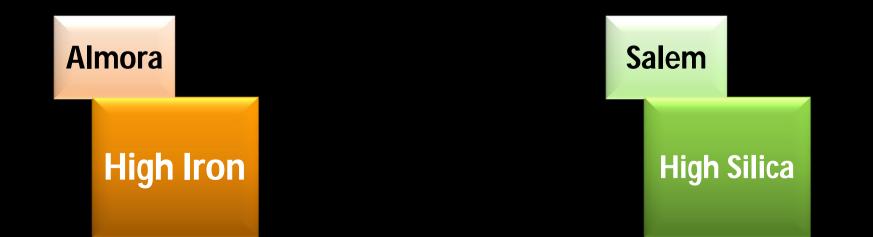
Antioxidants such as metallic powders are added mainly to suppress this oxidation.



Magnesia in Indian Refractories











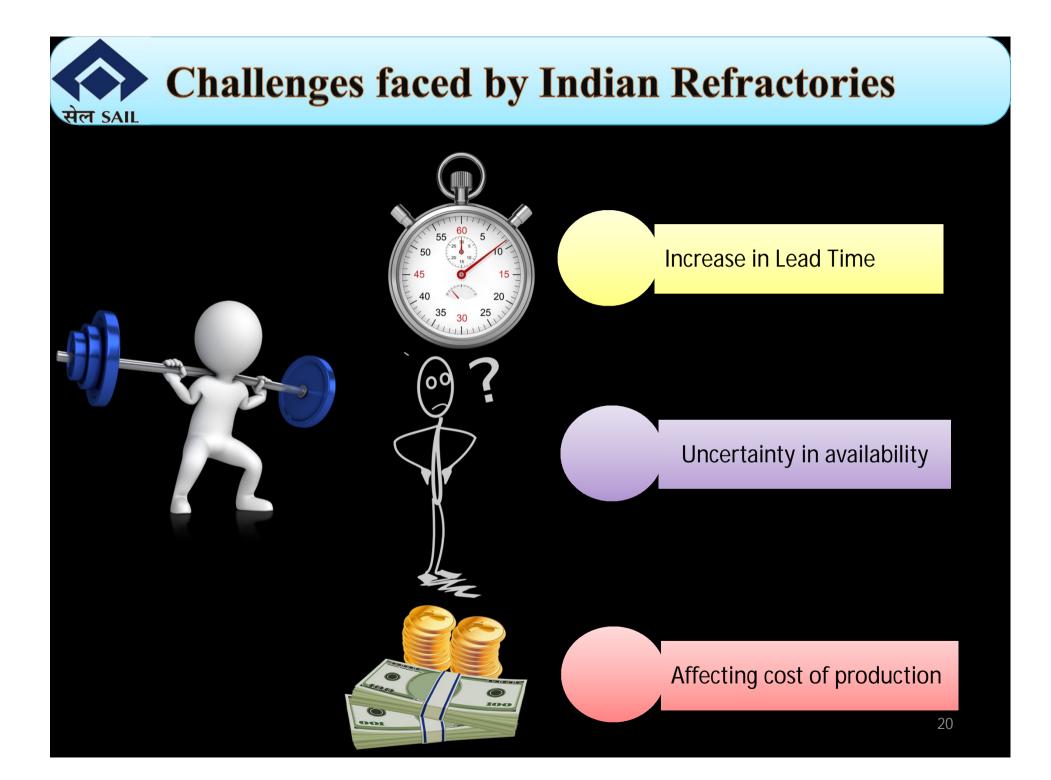
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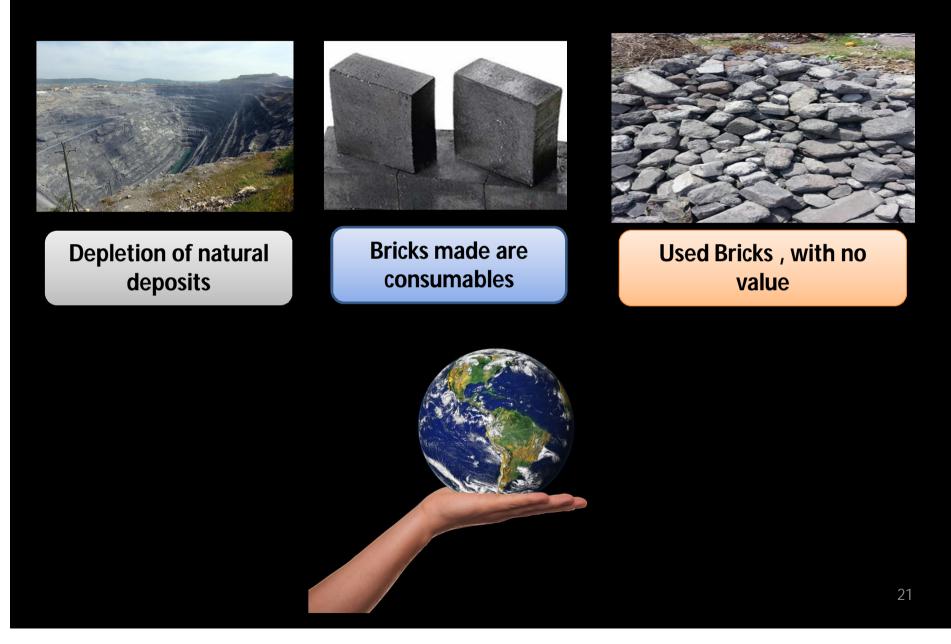
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Need of the hour







Thought Process

Resistance of graphite increases upto 3000°C. Hence the Carbon present in the leftover thickness should be unused.

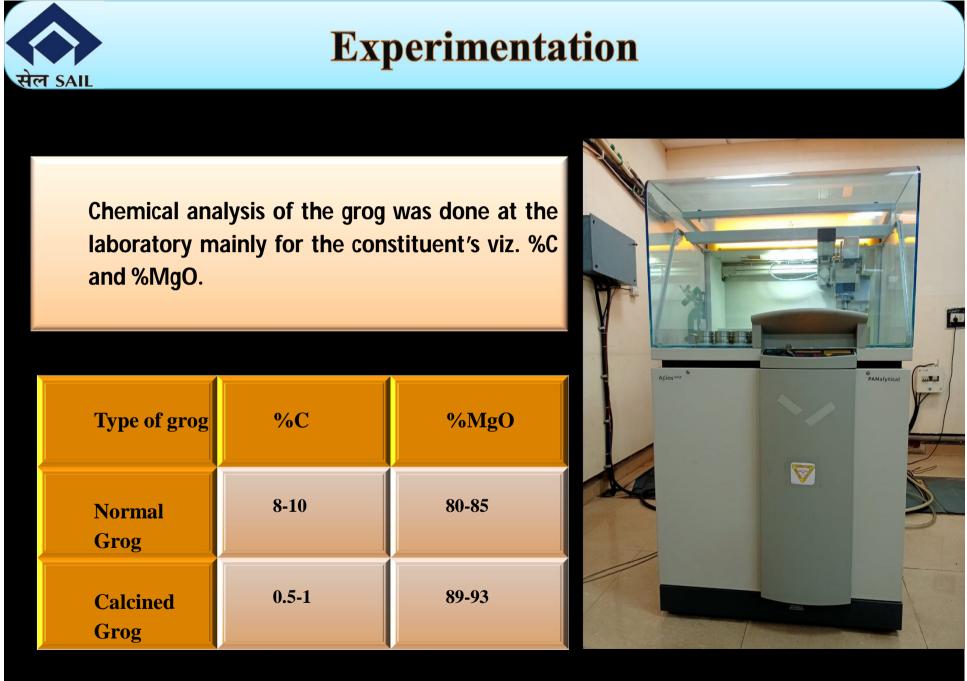
Melting point of magnesia is also high, hence the MgO present in the left over thickness remains untapped.

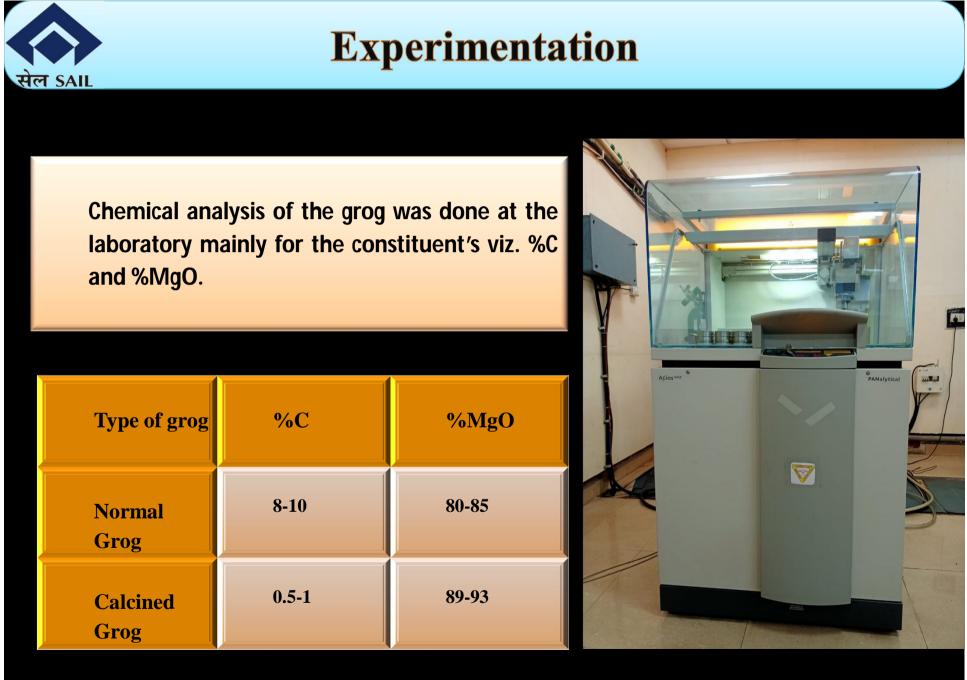
However the impurities such as Fe_2O_3 , CaO, SiO₂ increases due to the interaction with slag and metal

 Al_2O_3 from Aluminium powder combines with MgO at higher temperature to form Magnesium Aluminate spinel MgAl_2O_4.

Hence the presence of this spinel in grog may help us in getting better refractory properties.









Experimentation

Trial	АР	BD	CCS	COKE AP	COKE BD	COKE CCS	HMOR
C1	4.2	2.95	415	12.1	2.85	212.9	43.3
C2	4.3	2.92	350	13.7	2.98	237.1	65.4
C3	4.25	2.95	360	12.5	2.92	186.7	37.5
C4	4.3	2.93	315	12.1	2.88	220	58
C5	4.05	2.99	422	10.9	2.93	179.4	48.3
C6	4.35	2.94	322	10.4	2.91	187.3	61.9
C7	4.25	2.95	410	11.5	2.95	181.7	47.2
C8	4.05	3.0	387	11.4	2.85	205.2	78.7





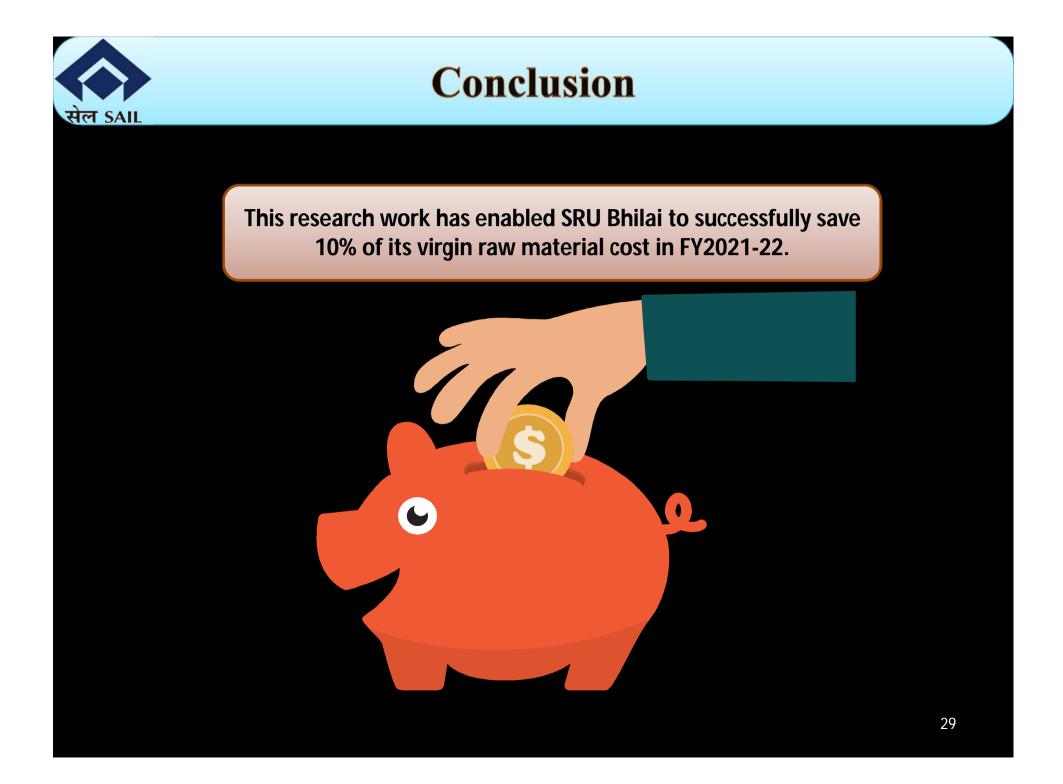


Conclusion

Used MgO-C brick, can be a partial substitute of virgin raw material for manufacturing of MgO-C bricks

Recycling of bricks has partially reduced our dependency on imported magnesia.

Improved the environmental sustainability which is the burning topic today.



Thank You